

NAME: PERIOD: DATE:

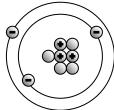
Atomic Facts Classwork

Answer all questions as best you can. Be specific. Be scientific. Use your periodic table to help.

- 1. What is the **atomic number** of an atom?
- 2. What is the mass number of an atom?

Look at the atoms below. Fill in their characteristics.

3.



atomic # -

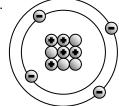
chemical symbol -

mass # -

isotope name -

+, -, or not an ion? -

4.



atomic#-

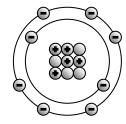
chemical symbol -

mass # -

isotope name -

+, -, or not an ion? -

5.



atomic # -

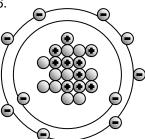
chemical symbol -

mass # -

isotope name -

+, -, or not an ion? -

6.



atomic # -

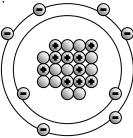
chemical symbol -

mass # -

isotope name -

+, -, or not an ion? -

7.



atomic # -

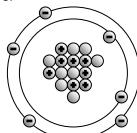
chemical symbol -

mass # -

isotope name -

+, -, or not an ion? -

8.



atomic#-

chemical symbol -

mass # -

isotope name -

+, -, or not an ion? -

9. Two of the above atoms (in 6-8) are isotopes. Which two atoms and why?